

# Microplastics Analysis in the Environment

Making it fast, simple, easy & efficient

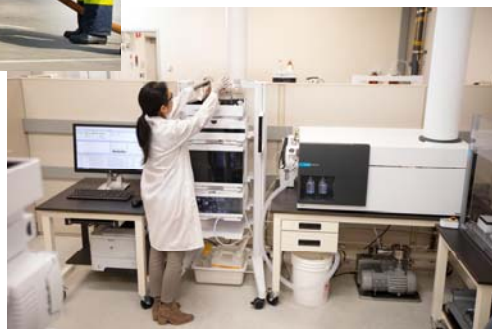
Tarun Anumol, Ph.D.  
Director, Global Environment & Food Markets  
[tarun.anumol@agilent.com](mailto:tarun.anumol@agilent.com)

Jeff Prevatt, Ph.D.  
Deputy Director, PIMA County Wastewater  
[Jeff.Prevatt@pima.gov](mailto:Jeff.Prevatt@pima.gov)

Microplastics NEMC Agilent Lunch Seminar



## Agilent at NEMC



### Leveraging Instrument Sensitivity – Customers, Laboratories, and Regulators Benefitting from Evolving Lab Practices

Brian Pike, Lead Chemist, IDEA Lab  
PACE Analytical

Virtual Lunch Seminar on Monday, August 17<sup>th</sup> will cover how the environmental testing industry offerings have been changing over time as a direct correlation to the sensitivity gains being realized through instrumentation for PFAS, Dioxins, sVOCs etc.



<https://www.agilent.com/en/product/liquid-chromatography-mass-spectrometry-lc-ms>

# We Don't Yet Know the Potential Health Impacts for Humans

## Global Health Organizations have Called for Additional Research

Given the potential impact, it is not surprising that global health organizations are also publishing reports on the impact of microplastics.

For example, in 2019, the World Health Organization (WHO) published [a report](#), which concluded that there was not enough scientific evidence regarding the impact of microplastics on human health and more scientific research is required.

"The lack of standard methods for sampling and analyzing microplastics in the environment means that comparisons across studies are difficult... To better assess human health risks and inform management actions, a number of research gaps need to be filled."

The World Health Organization, 2019.

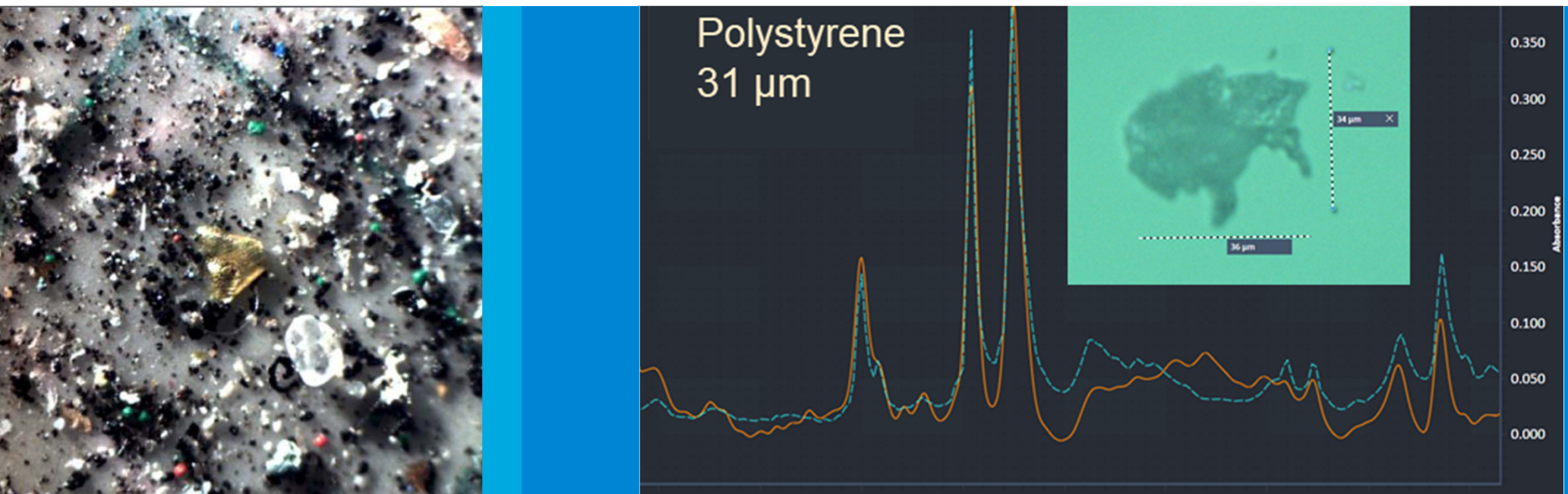
Microplastics in  
**drinking-**  
water



## Analysis of Microplastics A Challenge

Microplastics analysis require several different measurements depending on the goal and can include particle type, size, shape, area, & number

Hence there is no 'one' ideal solution for all the measurements but complementary techniques are required.



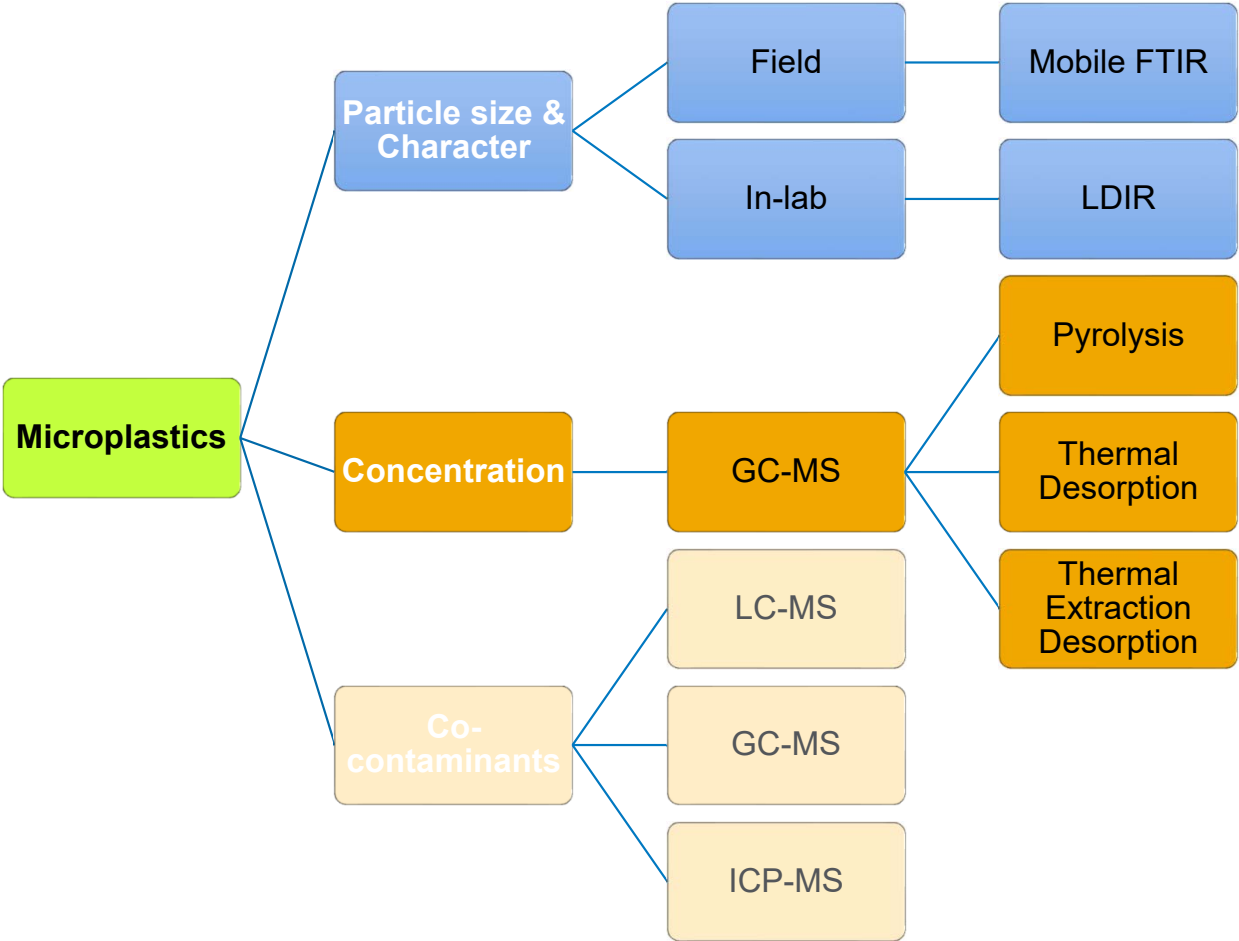
# Analysis of Microplastics in the Environment

Measurement technique selected depends upon question(s) being asked

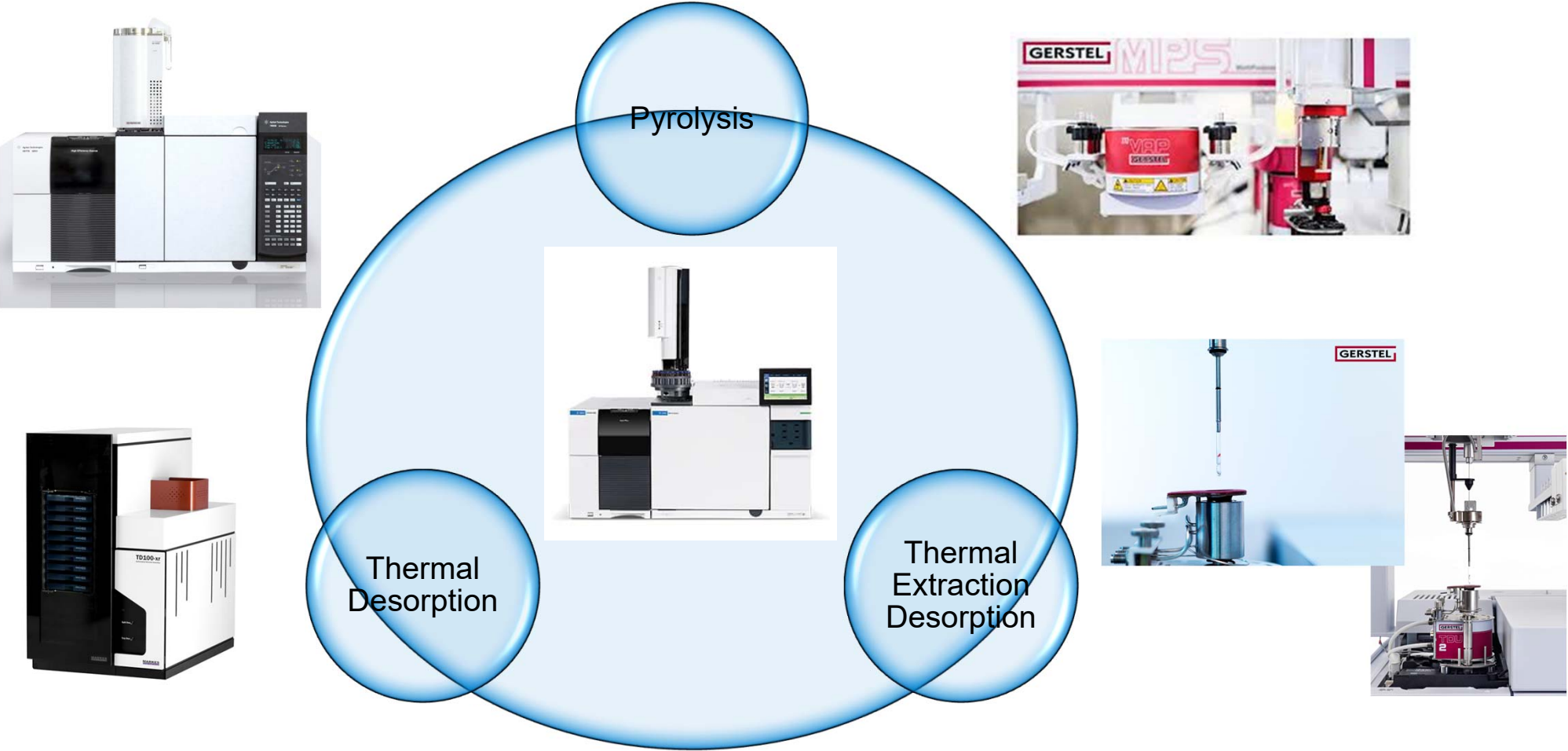
	Spectroscopic					Thermal	
	Optical/Fluorescence Microscopy	ATR-FTIR	$\mu$ FT-IR Imaging/LDIR	NIR	$\mu$ Raman	Pyr-GC/MS	T(E)D-GC/MS
Particle number	✓	✗	✓	✓	✓	✗	✗
ID	✗	✓	✓	✓	✓	✓*	✓*
Area/Shape	✓	✓	✓	✓	✓	✗	✗
Mass Fraction conc.	✗	✗	✗	✗	✗	✓	✓

\* Some types of plastics

# Microplastics Measurement



# Sample introduction approaches for mass fraction Microplastics analysis



## Thermogravimetric Analysis with Evolved Gas Analysis Why Pyrolysis?

---

Decomposition temperature area of environmental matrices: 100 – 600°C

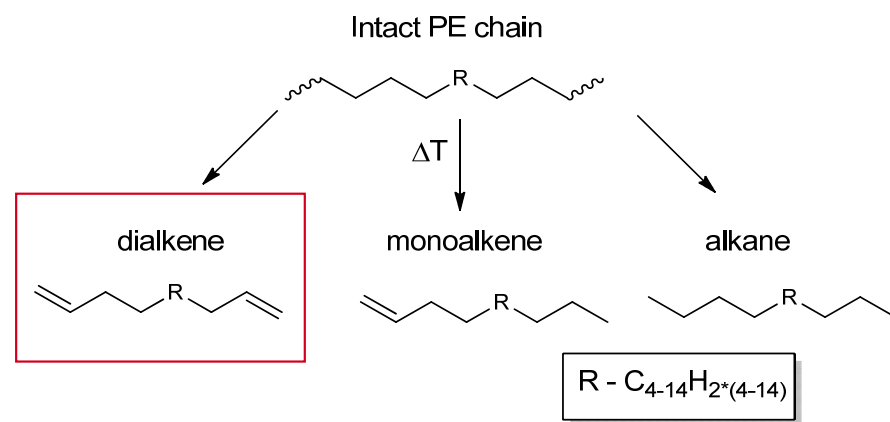
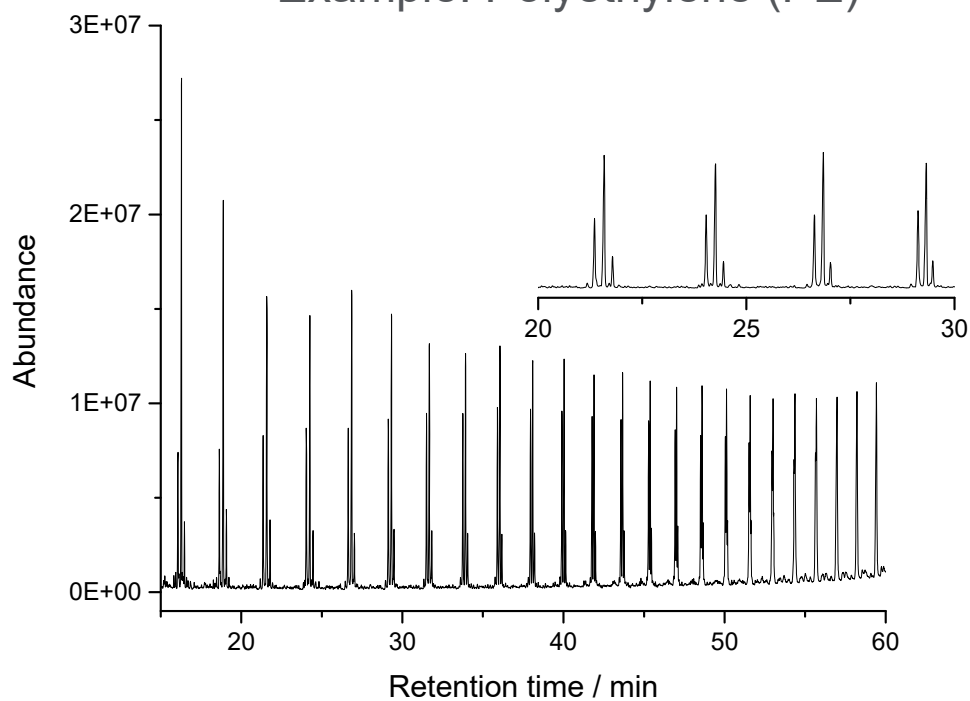
	<b>PP</b>	<b>PE</b>	<b>PET</b>	<b>PS</b>	<b>PA6</b>	<b>PVC</b>
Decomposition temperature area / °C	380 – 480	400 – 500	350 – 450	360 – 450	375 – 480	200 - 500
Gaseous decomposition products	methyl sub. alkanes + alkenes	alkenes, alkanes	CO <sub>2</sub> , ethene, arylic acids, esters	styrene, styrene derivates, oligomers	CO <sub>2</sub> , NH <sub>3</sub> , capro-lactam, amides	HCl, methane, alkene

 Chromatographic separation of decomposition products is necessary



# Pyrolysis-Gas Chromatography-Mass Spectrometry (Py-GC-MS)

## Example: Polyethylene (PE)



➔ Certain identification is possible, but only with small sample mass (~ max. 1 mg or organic substances)

# Pyrolysis GC/MS for Microplastics

## Several Agilent users performing this analysis

Application Note  
Environmental



### Quantification of Microplastics in Environmental Samples Using Pyrolysis and GC/MSD

#### Authors

George Dierkes,  
Susanne Becher,  
Heike Schumacher, and  
Corinna Földi  
German Federal Institute of  
Hydrology, Koblenz, Germany  
Tim Lauschke, and  
Thomas Ternes  
Institute of Integrated  
Natural Sciences, University  
Koblenz-Landau, Germany  
Joerg Riener  
Agilent Technologies, Inc.

#### Abstract

There is growing interest in quantifying microplastics in environmental samples. This application note presents a robust pressurized liquid extraction (PLE) with pyrolysis-gas chromatography-mass spectrometry (pyr-GC/MS) method for quantitation of microplastics like polyethylene (PE), polypropylene (PP), and polystyrene (PS) at low concentrations in environmental matrices using the Agilent 5977B GC/MSD, Agilent 7890B GC, and Agilent MassHunter workstation software. Linearity, limits of quantitation (LOQs), and reproducibility for real environmental samples were evaluated. The GC/MSD addressed the insufficient limits of detection that have challenged previous methods. PE, PP, and PS microplastics were quantified down to 0.005 mg/g. Excellent linearity ( $R^2 > 0.97$ ) for calibration samples from 0.005 to 1 mg/g was obtained. Relative standard deviations (RSDs) for both spiked and environmental samples were <10% or lower, demonstrating excellent system reproducibility and reliability.

Agilent Application: 5994-2199EN

Analytical and Bioanalytical Chemistry (2019) 41:1785–1790  
<https://doi.org/10.1007/s00216-019-02066-9>

RESEARCH PAPER



### Quantification of microplastics in environmental samples via pressurized liquid extraction and pyrolysis-gas chromatography

George Dierkes<sup>1</sup> · Tim Lauschke<sup>1,2</sup> · Susanne Becher<sup>1</sup> · Heike Schumacher<sup>1</sup> · Corinna Földi<sup>1</sup> · Thomas Ternes<sup>1,2</sup>

Received: 14 June 2019 / Revised: 15 July 2019 / Accepted: 1 August 2019 / Published online: 30 August 2019  
© Springer-Verlag GmbH Germany, part of Springer Nature 2019

ENVIRONMENTAL  
Science & Technology

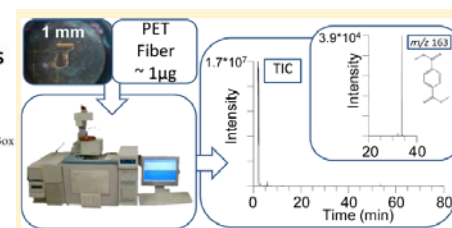
Article  
pubs.acs.org/est

### Simultaneous Trace Identification and Quantification of Common Types of Microplastics in Environmental Samples by Pyrolysis-Gas Chromatography–Mass Spectrometry

Marten Fischer and Barbara M. Scholz-Böttcher<sup>\*†</sup>

Institute for Chemistry and Biology of the Marine Environment (ICBM), Carl von Ossietzky University of Oldenburg, P.O. Box 2503, D-26111 Oldenburg, Germany

PE, PP, PS,  
PVC, PET,  
PC, PMMA,  
PA6



analytical  
chemistry

Cite This: Anal. Chem. 2019, 91, 1785–1790

Technical Note  
pubs.acs.org/ac

### Cloud-Point Extraction Combined with Thermal Degradation for Nanoplastic Analysis Using Pyrolysis Gas Chromatography–Mass Spectrometry

Xiao-xia Zhou,<sup>†,‡</sup> Li-teng Hao,<sup>‡</sup> Huang-ying-zi Wang,<sup>†</sup> Ying-jie Li,<sup>‡</sup> and Jing-fu Liu<sup>\*‡</sup>

<sup>†</sup>Key Laboratory for Water Quality and Conservation of the Pearl River Delta, Ministry of Education, Institute of Environmental Research at Greater Bay, Guangzhou University, Guangzhou 510006, China

<sup>‡</sup>State Key Laboratory of Environmental Chemistry and Ecotoxicology, Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences, P.O. Box 2871, Beijing 100085, China

# Thermal Extraction-Desorption Gas Chromatography Mass Spectrometry. TED-GC/MS



## Components of TED-GC/MS

- Agilent 8890 GC
- Agilent 5977B GC/MSD
- Gerstel TDU
- Gerstel Twister
- Mettler Toledo TGA

## Advantages:

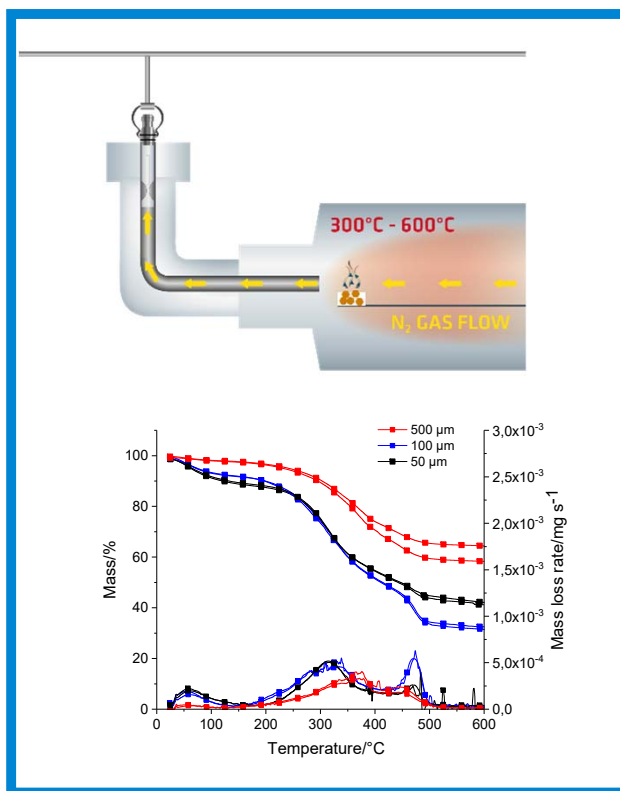
Fully automated including sample prep.  
Homogeneity control with TGA  
Better sample cleanup  
Higher sample loads  
More sensitivity

# TED-GC-MS: 1st Part: Thermoextraction



Environmental sample =  
matrix + microplastics

## Thermobalance



Adsorber loaded with  
decomposition products

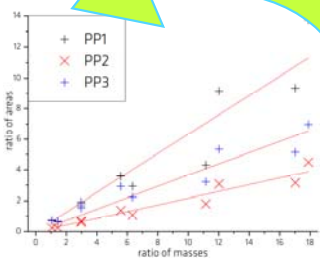
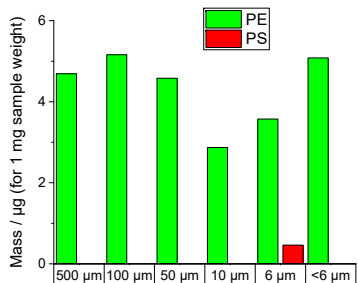
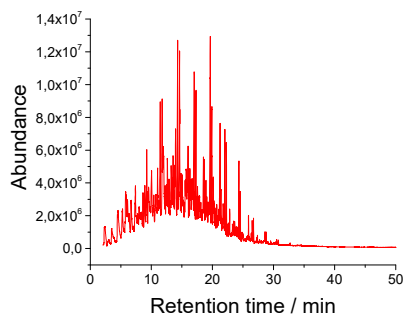
Pyrolysis: inert atmosphere

RT – 600 °C: Pyrolytic  
decomposable organic compounds

TGA signal: Homogeneity control

# TED-GC-MS

## 2nd Part: Analysis of Decomposition Products



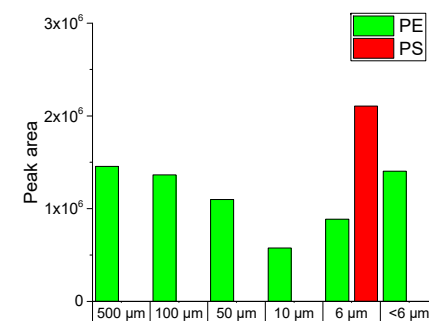
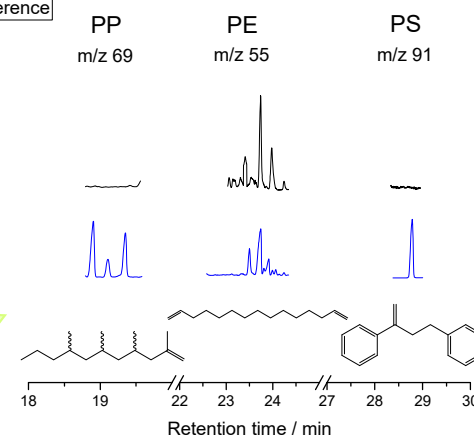
Quantification of polymers

Thermal Desorption GC-MS

Identification of marker molecules

Identification of polymers

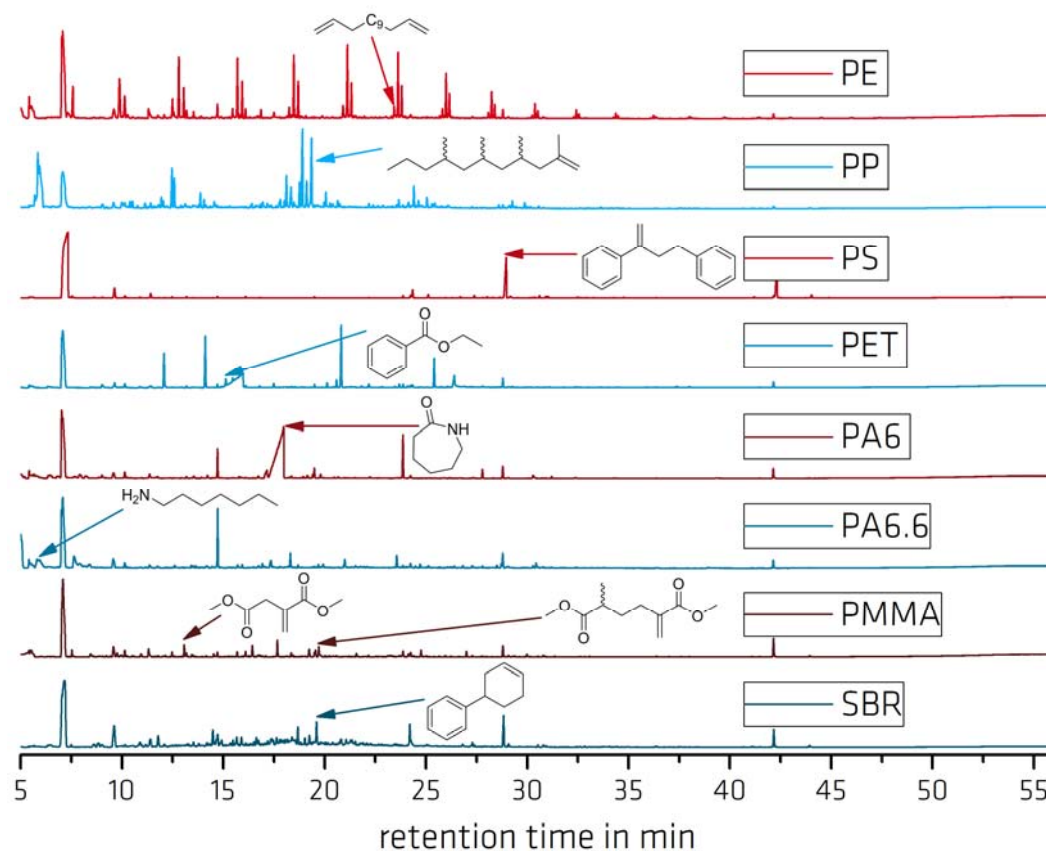
Sample  
Reference



# TED-GC-MS

## specific decomposition products of polymers

- PE, PP, PS
- PET, PA, PMMA
- PBAT, PLA
- SBR = Styrene-Butadiene-Rubber (component of tyres)
- LOD: between 0.08 µg (PS) and 2.2 µg (PE)



# Thermal Extraction Desorption-Gas chromatography-Mass Spectrometry (TED-GC-MS)

Water Research 85 (2015) 451–457

Contents lists available at ScienceDirect

## Water Research

journal homepage: [www.elsevier.com/locate/watres](http://www.elsevier.com/locate/watres)

### Analysis of polyethylene microplastics in environmental samples, using a thermal decomposition method

Erik Dümichen<sup>a</sup>, Anne-Kathrin Barthel<sup>a</sup>, Ulrike Braun<sup>a,\*,</sup>, Claus G. Bannick<sup>b</sup>, Kathrin Brand<sup>b,c</sup>, Martin Jekel<sup>c</sup>, Rainer Senz<sup>d</sup>

<sup>a</sup> BAM Federal Institute for Material Research and Testing, Unter den Eichen 87, 12205 Berlin, Germany  
<sup>b</sup> UBA Umweltbundesamt, Wörlitzer Platz 1, 06844 Dessau-Roßlau, Germany  
<sup>c</sup> Technical University of Berlin, Müser Urban Am  
<sup>d</sup> Beuth University of Applied Sciences, Leumnitz

Environmental Pollution 231 (2017) 1236–1264

Contents lists available at ScienceDirect

## Environmental Pollution

journal homepage: [www.elsevier.com/locate/envpol](http://www.elsevier.com/locate/envpol)

### Comparison of different methods for MP detection: What can we learn from them, and why asking the right question before measurements matters?<sup>☆</sup>

Anna M. Elert<sup>a</sup>, Roland Becker, Erik Duemichen, Paul Eisentraut, Jana Falkenhagen, Heinz Sturm, Ulrike Braun

<sup>a</sup> Federal Institute for Material Research and Testing (BAM), Unter den Eichen 87, 12205 Berlin, Germany

May 29, 2018

## Polymer Analysis and Microplastics in the Environment

### TED-GC-MS, a New and Fast Method

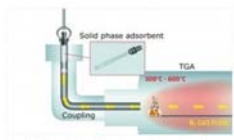


Fig. 1. Schematic of the thermal extraction (pyrolytic decomposition) of a sample in the TGA and transport of the resulting decomposition fragments to a solid-phase adsorbent, which is positioned in a thermal desorption glass tube located in a newly developed coupling device.

The abbreviation TED-GC-MS describes a two-step method for volatile diagnostics

ARTICLE INFO

Article history:  
Received 2 June 2015  
Received in revised form 28 August 2015  
Accepted 3 September 2015

- Chemistry & Physics
- Pharma & Drug Discovery
- Life Sciences & Biotech
- Food
- Environment
- Medicine & Diagnostics
- Material Science
- Information Technology (IT)
- Literature

Journal of Chromatography A, 1592 (2019) 133–142

Contents lists available at ScienceDirect

## Journal of Chromatography A

journal homepage: [www.elsevier.com/locate/chroma](http://www.elsevier.com/locate/chroma)

### Automated thermal extraction-desorption gas chromatography mass spectrometry: A multifunctional tool for comprehensive characterization of polymers and their degradation products

E. Duemichen<sup>a,\*</sup>, P. Eisentraut<sup>a</sup>, M. Celina<sup>b</sup>, U. Braun<sup>a</sup>

<sup>a</sup> Bundesanstalt für Materialforschung und -prüfung (BAM), Unter den Eichen 87, 12205 Berlin, Germany  
<sup>b</sup> Sandia National Laboratories, Organic Materials Science Dept., 1853, Albuquerque, NM, 87185-1411, USA

ARTICLE INFO

Article history:  
Received 28 September 2018  
Received in revised form 8 January 2019  
Accepted 10 January 2019

ABSTRACT

The TED-GC-MS analysis is a two-step method. A sample is first decomposed in a thermogravimetric analyzer (TGA) and the gaseous decomposition products are then trapped on a solid-phase adsorbent. Subsequently, the solid-phase adsorbent is analyzed with thermal desorption gas chromatography mass

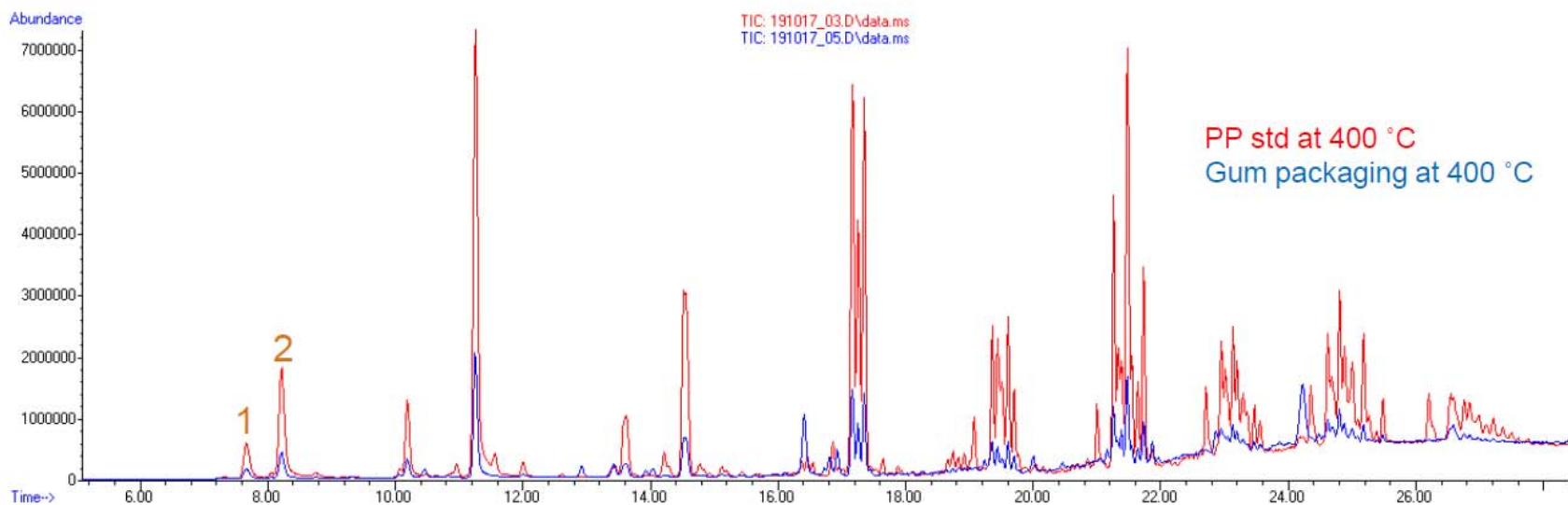
## Unique markers via TD

Polypropylene (PP) real plastics, the whole profile

- PP std
- Gum packaging

Slide Courtesy: Markes Intl.  
“Sampling & Analysis of Emerging  
Pollutants”, Hannah Calder

- Unique Compounds:
  1. tert-butanol  
(7.7 min, 59 ion),
  2. 2-methyl pentane  
(8 min, 43/57/71 ions)





# LC/MS/MS for specific plastics analysis

## A Simple Method for Quantifying Polycarbonate and Polyethylene Terephthalate Microplastics in Environmental Samples by Liquid Chromatography–Tandem Mass Spectrometry

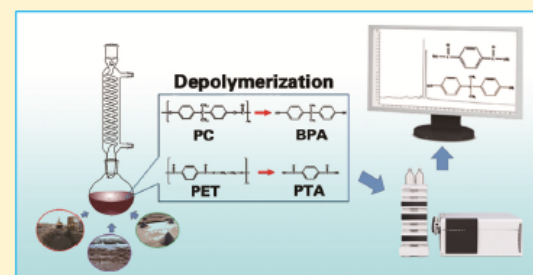
Lei Wang,<sup>\*†</sup> Junjie Zhang, Shaogang Hou, and Hongwen Sun<sup>‡</sup>

Ministry of Education Key Laboratory of Pollution Processes and Environmental Criteria, Tianjin Key Laboratory of Environmental Remediation and Pollution Control, College of Environmental Science and Engineering, Nankai University, Tianjin 300350, China

<sup>†</sup> Supporting Information



**ABSTRACT:** Microplastics (MPs) have frequently been found in the environment. However, studies of the quantification methods for MPs are still needed. Plastics are polymers with different degrees of polymerization. In this study, alkali-assisted thermal hydrolysis was applied to depolymerize two plastics containing ester groups, polycarbonate (PC) and polyethylene terephthalate (PET), in a pentanol or butanol system. By determining the concentrations of the depolymerized building block compounds, i.e., bisphenol A and *p*-phthalic acid, we quantified the amounts of PC and PET MPs in environmental samples. Recoveries of 87.2–97.1% were obtained for the PC and PET plastic particles spiked in the landfill sludge. The method was successfully applied to determine the occurrence of PC and PET MPs in samples of sludge, marine sediments, indoor dust, digestive residues in mussel and clam, and sea salt and rock salt. High concentrations of 246 and 430 mg/kg were determined for PC and PET type MPs, respectively, in an indoor dust. In addition, concentrations of 63.7 mg/kg for PC and 127 mg/kg for PET were detected in the digestive residues of a clam.



# Analysis of emerging contaminants sorbed to Microplastics

ENVIRONMENTAL  
Science & Technology

Article  
pubs.acs.org/est

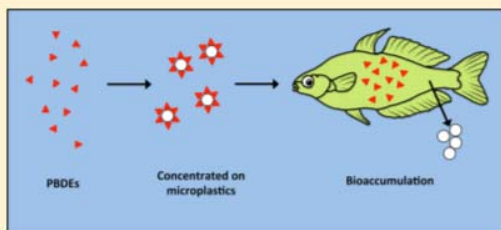
## Chemical Pollutants Sorbed to Ingested Microbeads from Personal Care Products Accumulate in Fish

Peter Wardrop,<sup>†</sup> Jeff Shimeta,<sup>†</sup> Dayanthi Nugegoda,<sup>†</sup> Paul D. Morrison,<sup>†</sup> Ana Miranda,<sup>‡</sup> Min Tang,<sup>‡</sup> and Bradley O. Clarke<sup>\*†</sup>

<sup>†</sup>Centre for Environmental Sustainability and Remediation, RMIT University, GPO Box 2476, Melbourne, Victoria 3001, Australia

<sup>‡</sup>Key Laboratory of Advanced Materials of Tropical Island Resources, Ministry of Education; School of Materials and Chemical Engineering, Hainan University, Haikou, Hainan 570228, China

Supporting Information



**ABSTRACT:** The prevalence of microplastics (<5 mm) in natural environments has become a widely recognized global problem. Microplastics have been shown to sorb chemical pollutants from their surrounding environment, thus raising concern as to their role in the movement of these pollutants through the food chain. This experiment investigated whether organic pollutants sorbed to microbeads (MBs) from personal care products were assimilated by fish following particle ingestion. Rainbow fish (*Melanotaenia fluviatilis*) were exposed to MBs with sorbed polybrominated diphenyl ethers (PBDEs; BDE-28, -47, -100, -99, -153, -154, -183, 200 ng g<sup>-1</sup>; BDE-209, 2000 ng g<sup>-1</sup>) and sampled at 0, 21, 42, and 63 days along with two control treatments (food only and food + clean MBs). Exposed fish had significantly higher Σ<sub>6</sub>PBDE concentrations than both control treatments after just 21 days, and continued exposure resulted in increased accumulation of the pollutants over the experiment (ca. 115 pg g<sup>-1</sup> ww d<sup>-1</sup>). Lower brominated congeners showed the highest assimilation whereas higher brominated congeners did not appear to transfer, indicating they may be too strongly sorbed to the plastic or unable to be assimilated by the fish due to large molecular size or other factors. Seemingly against this trend, however, BDE-99 did not appear to bioaccumulate in the fish, which may be due to partitioning from the MBs or it being metabolized in vivo. This work provides evidence that MBs from personal care products are capable of transferring sorbed pollutants to fish that ingest them.



Bradley Clarke  
Melbourne University



7000 Series GC/TQ

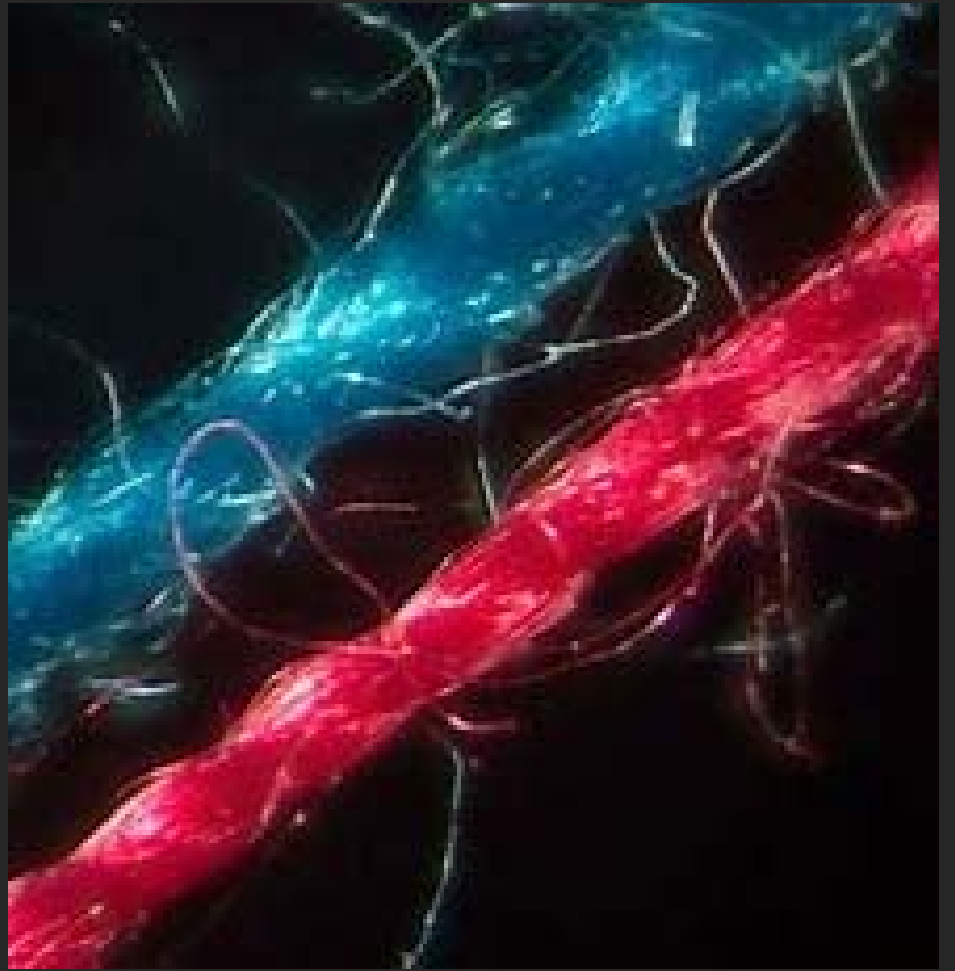
<https://doi.org/10.1021/acs.est.5b06280>

# Microplastics Analysis:

Simplified Workflows for  
Comprehensive Characterization  
in the Environment

Jeff Prevatt, PhD







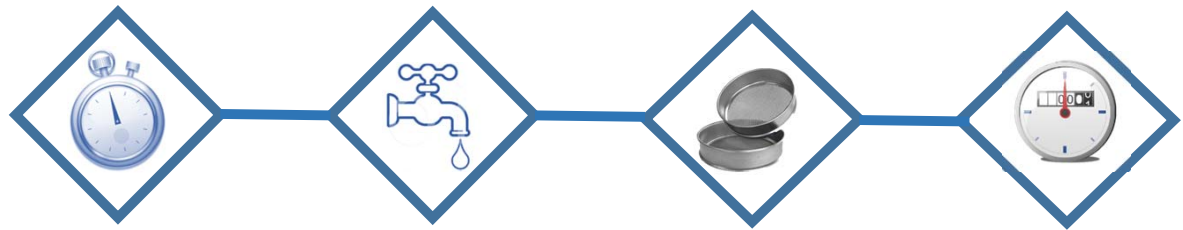
# Workflow Development

---

- Ease of use
- Sample preparation
- Speed of analysis
- Data interpretation
- Reporting



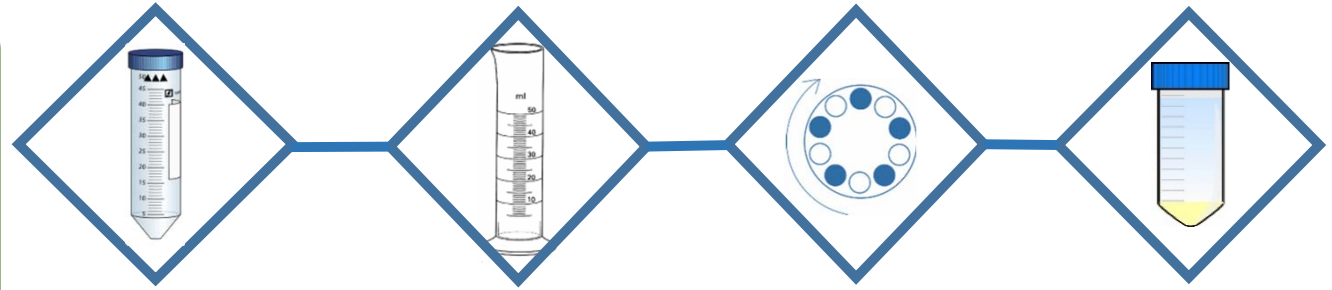
## Sample Collection Workflow



1. Straightforward
2. Simple to follow
3. Reproducible

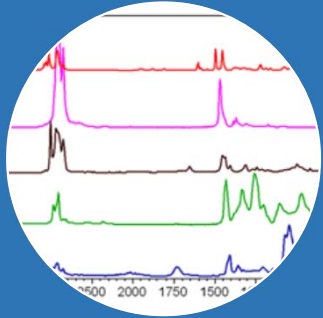


## Sample Preparation Workflow



1. Easy to replicate
2. Addresses Interferences
3. Minimizes loss of particles





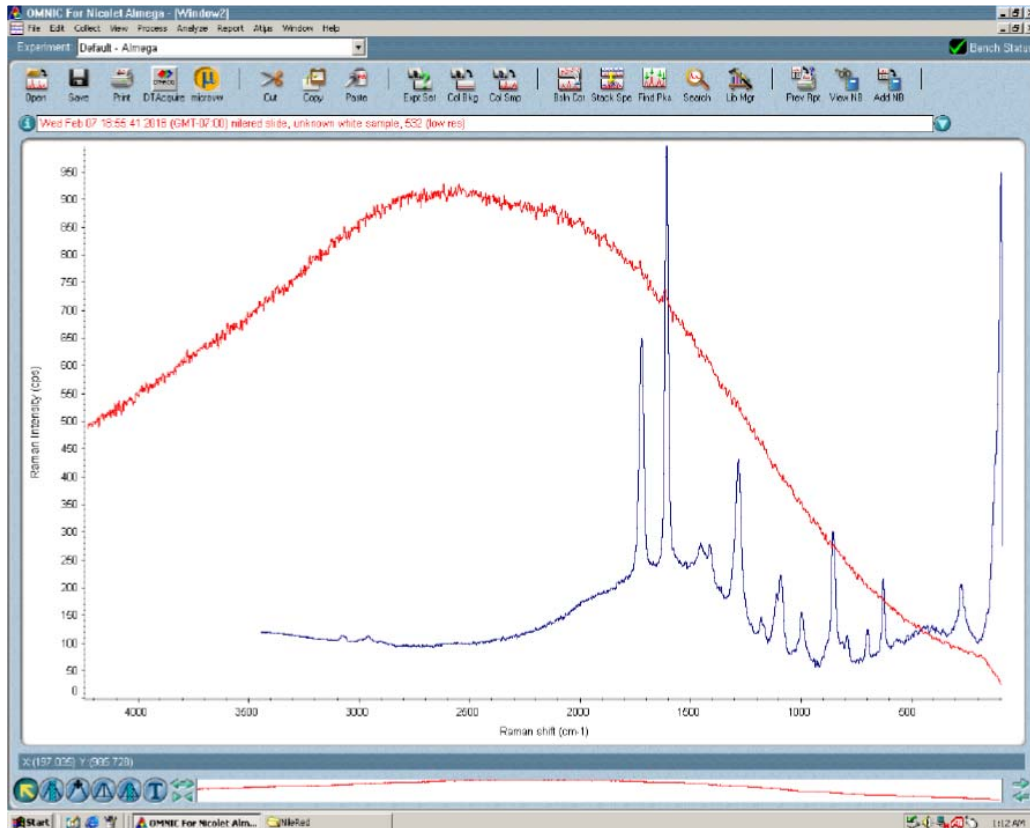
## Sample Analysis Workflow



1. Identification of particles
2. Quantification of particles
3. Particle size distribution
4. Fast & reproducible

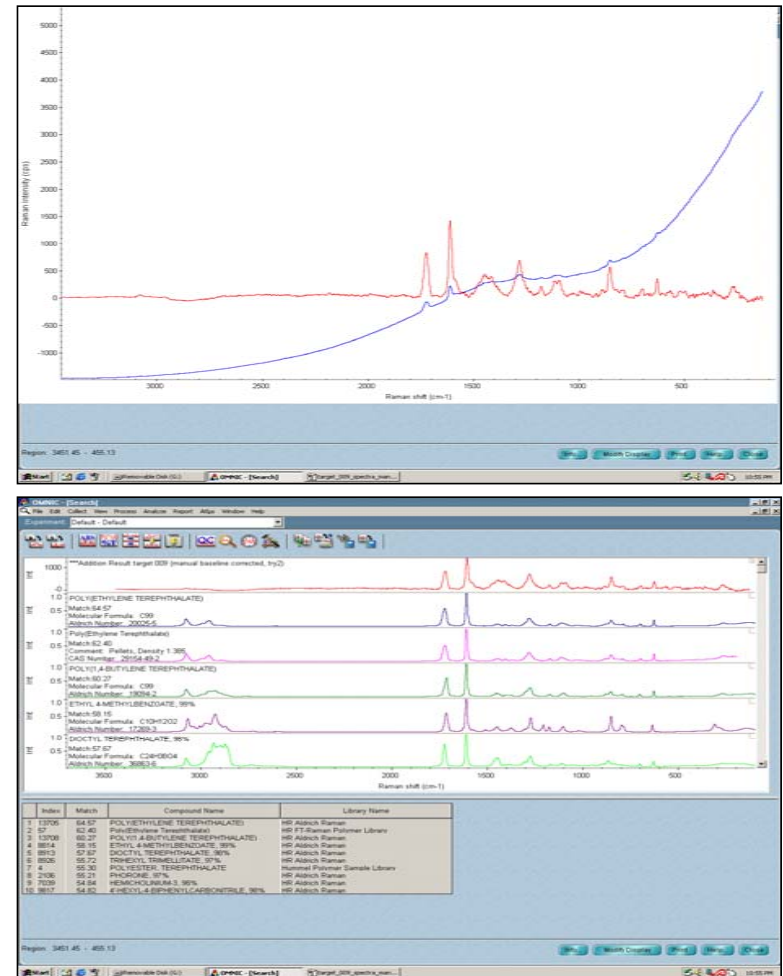
# Micro-Raman Analysis

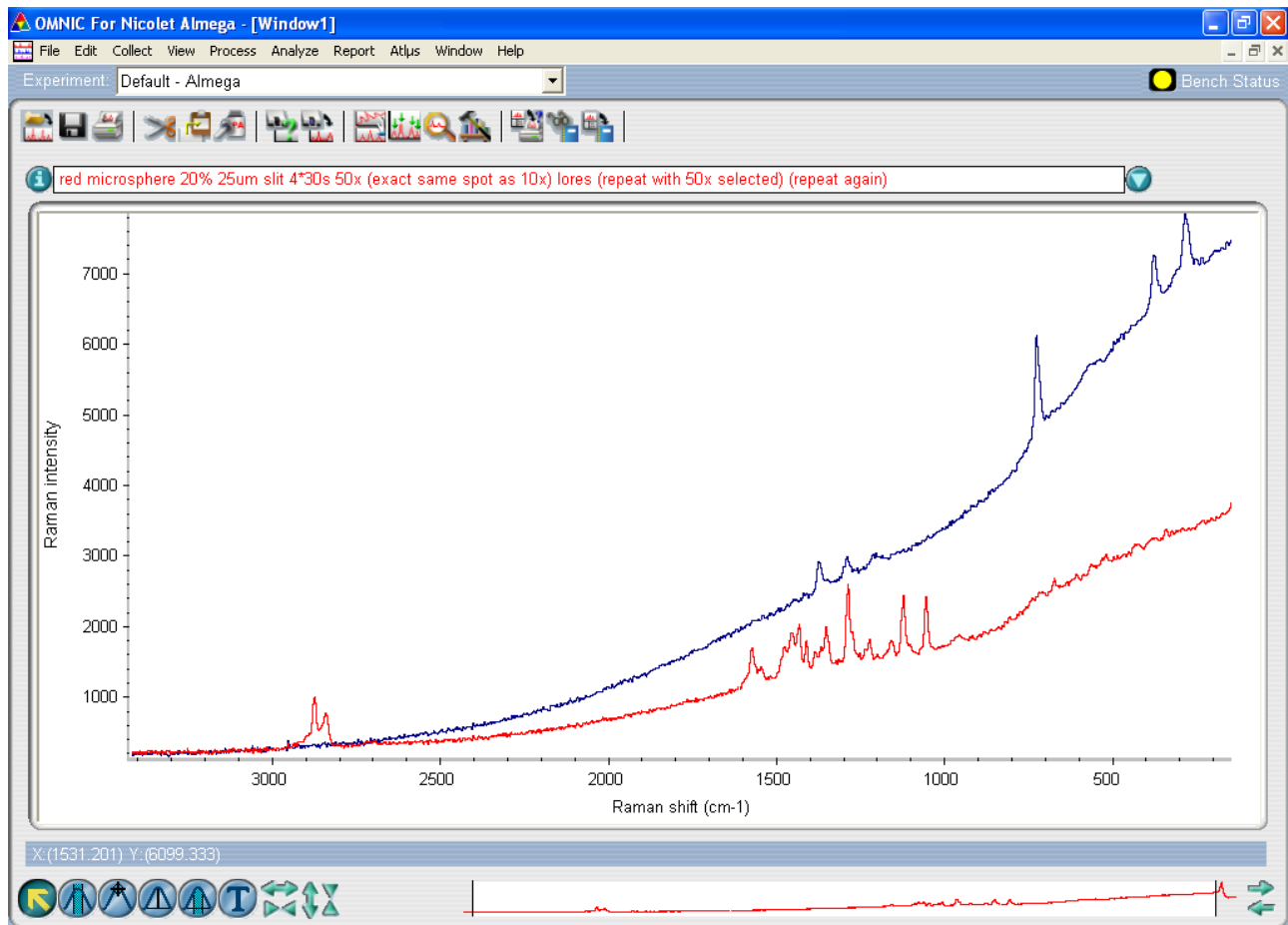
---



- ✓ FTIR and Raman spectroscopy are suitable for definitive identification of plastics
- ✗ 532 nm excitation source results in excessive fluorescence
- ✓ 785 nm excitation source produces acceptable signal for differentiating plastics

# Polyethylene





# Challenges

Instrument setting vary

Analysis times are slow

Multiple acquisitions required

Samples easily damaged

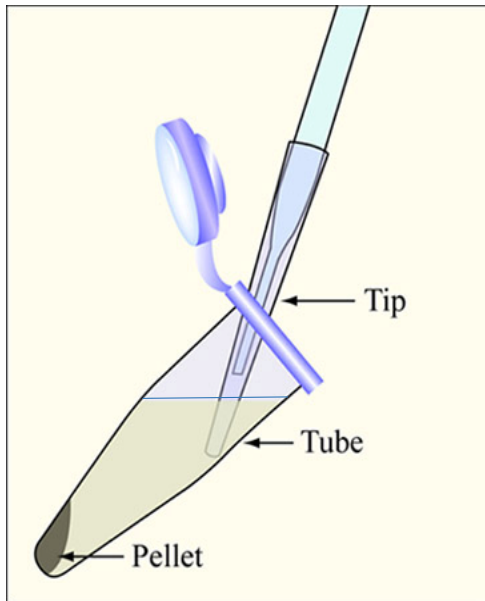


Agilent  
LDIR 8700

Chemical  
Imaging  
System

# Agilent LDIR 8700

Specification	Description
Optical Resolution	1 micron
Spectral Resolution	20 micron
Wavelength Coverage	1,800 – 1,000 wavenumbers (cm <sup>-1</sup> )
Light Source	Quantum Cascade Laser
Detector	TE cooled MCT
Time Required for Analysis	Minutes to hours
Reference Library	IR reference and buildable



- Eliminates filtration and associated sample loss
- Cellulose removed
- Preparation time is ideal
- Sieved contents plated onto a slide
- Reference spike recovery near 100%

# Particle Analysis

Library Microplastics Lib

Particles

Identifications

Statistics

Settings

Auto Scan

Particle Sensitivity



Particle Diameter ( $\mu\text{m}$ )

Minimum

100



Auto

Maximum

5000



Auto







Library Microplastics Lib

- Particles
- Identifications
- Statistics
- Settings

Highlight particles on image

<input checked="" type="checkbox"/> Unknown	<div style="width: 50%; height: 10px; background-color: #007bff;"></div>	50.0% (10)
<input checked="" type="checkbox"/> Cellulosic	<div style="width: 15%; height: 10px; background-color: #90ee90;"></div>	15.0% (3)
<input checked="" type="checkbox"/> Silica	<div style="width: 10%; height: 10px; background-color: #add8e6;"></div>	10.0% (2)
<input checked="" type="checkbox"/> Polyurethane (PU)	<div style="width: 10%; height: 10px; background-color: #ff8c00;"></div>	10.0% (2)
<input checked="" type="checkbox"/> Chitin	<div style="width: 10%; height: 10px; background-color: #ff7f7f;"></div>	10.0% (2)
<input checked="" type="checkbox"/> Calcium Stearate	<div style="width: 5%; height: 10px; background-color: #add8e6;"></div>	5.0% (1)

## Particle Analysis

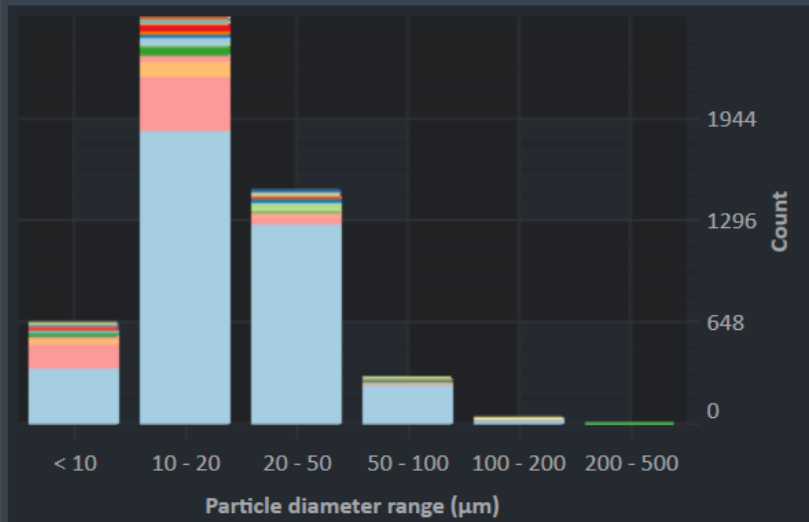
Library Microplastics Starter 1.0

Particles Identifications Statistics Settings

Highlight particles on image

<input checked="" type="checkbox"/> PP	75.1% (3791)
<input checked="" type="checkbox"/> Chitin	10.8% (543)
<input checked="" type="checkbox"/> Polysulfones	2.7% (136)
<input checked="" type="checkbox"/> Cellulosic	1.7% (87)
<input checked="" type="checkbox"/> Polyether	1.6% (82)
<input checked="" type="checkbox"/> Polyamide (PA)	1.3% (68)
<input checked="" type="checkbox"/> Coal	1.3% (65)
<input checked="" type="checkbox"/> Silica	1.3% (64)
<input checked="" type="checkbox"/> Natural Polyamide	0.8% (40)
<input checked="" type="checkbox"/> Rubber	0.8% (39)
<input checked="" type="checkbox"/> Polyethylene Terephthalate (PET)	0.7% (34)

## Group Sizes



- PP
- Chitin
- Polysulfones
- Cellulosic
- Polyether
- Polyamide (PA)
- Coal
- Silica
- Natural Polyamide
- Rubber



## Particle Analysis

Library Microplastics Lib

Particles

Identifications

Statistics

Settings

Auto Scan

Particle Sensitivity

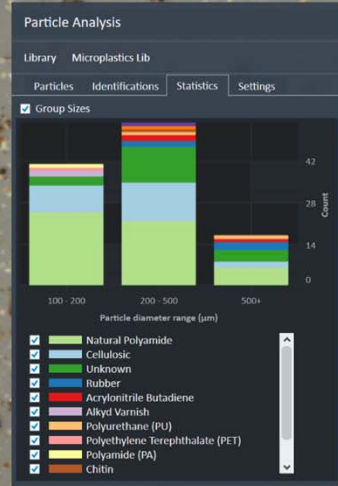


Particle Diameter ( $\mu\text{m}$ )

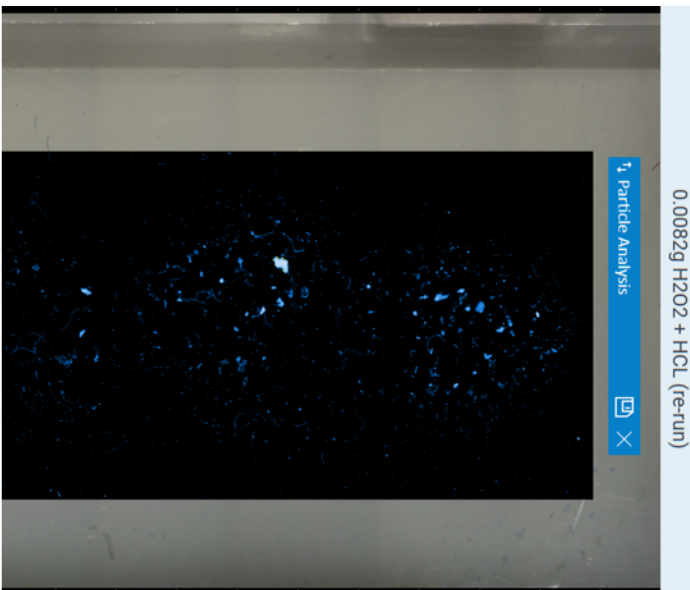
Minimum   Auto

Maximum   Auto

Particle Analysis



1 mm



0.0082g H2O2 + HCl (re-run)

### Particle Analysis

Library Microplastics Starter 1.0

Particles Identifications Statistics **Settings**

Auto Scan

Collect Visible Image

Particle Sensitivity

Classification Ranges

Particle Diameter (µm)

Minimum   Auto

Maximum   Auto

### Particle Analysis

Library Microplastics Starter 1.0

Particles Identifications **Statistics** Settings

Group Sizes

Particle diameter range (µm)

- PP
- Polyurethane (PU)
- Natural Polyamide
- Acrylates
- Alkyd Varnish
- Polyethylene Terephthalate (PET)
- Cellulosic
- Rubber
- Polyamide (PA)
- Polyethylene (PE)

### Particle Analysis

Library Microplastics Starter 1.0

Particles Identifications **Statistics** Settings

Highlight particles on image

- PP 94.5% (2322)
- Polyurethane (PU) 1.7% (43)
- Natural Polyamide 1.3% (31)
- Acrylates 0.7% (16)
- Alkyd Varnish 0.6% (14)
- Polyethylene Terephthalate (PET)

View All 2,458 particles

Search by Id

pp

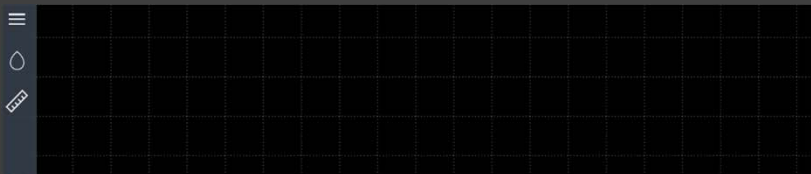
Quality 0.858

Id # A3  Accept Prediction

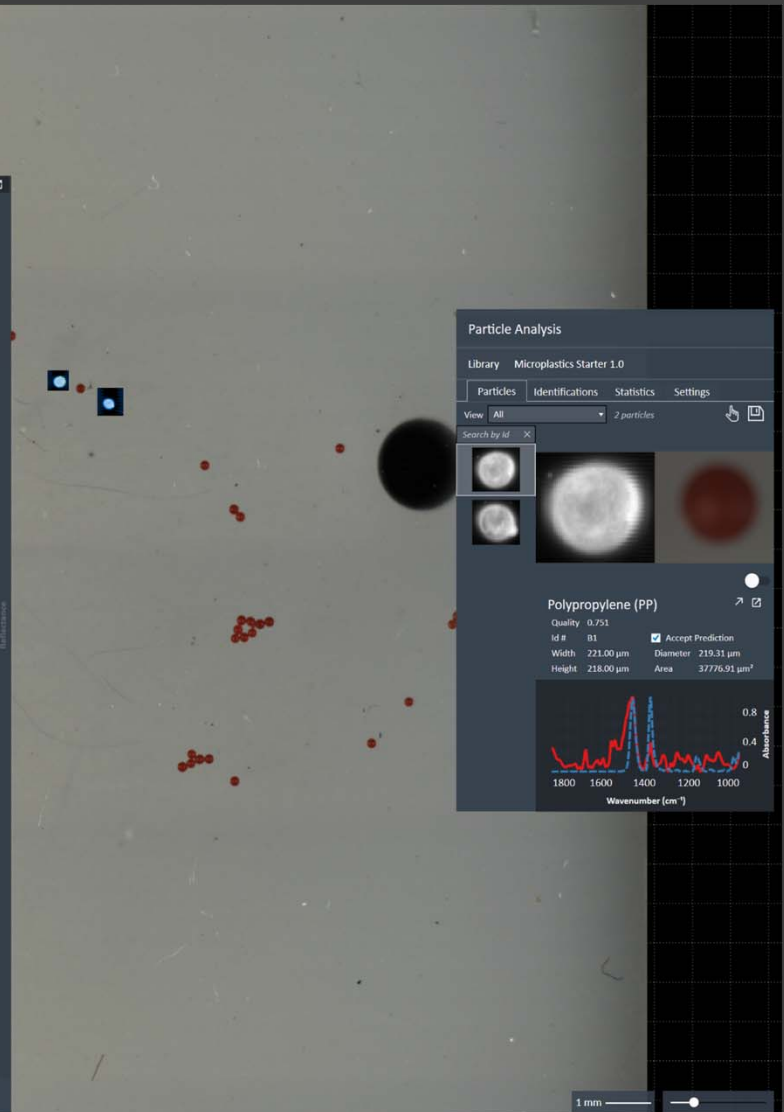
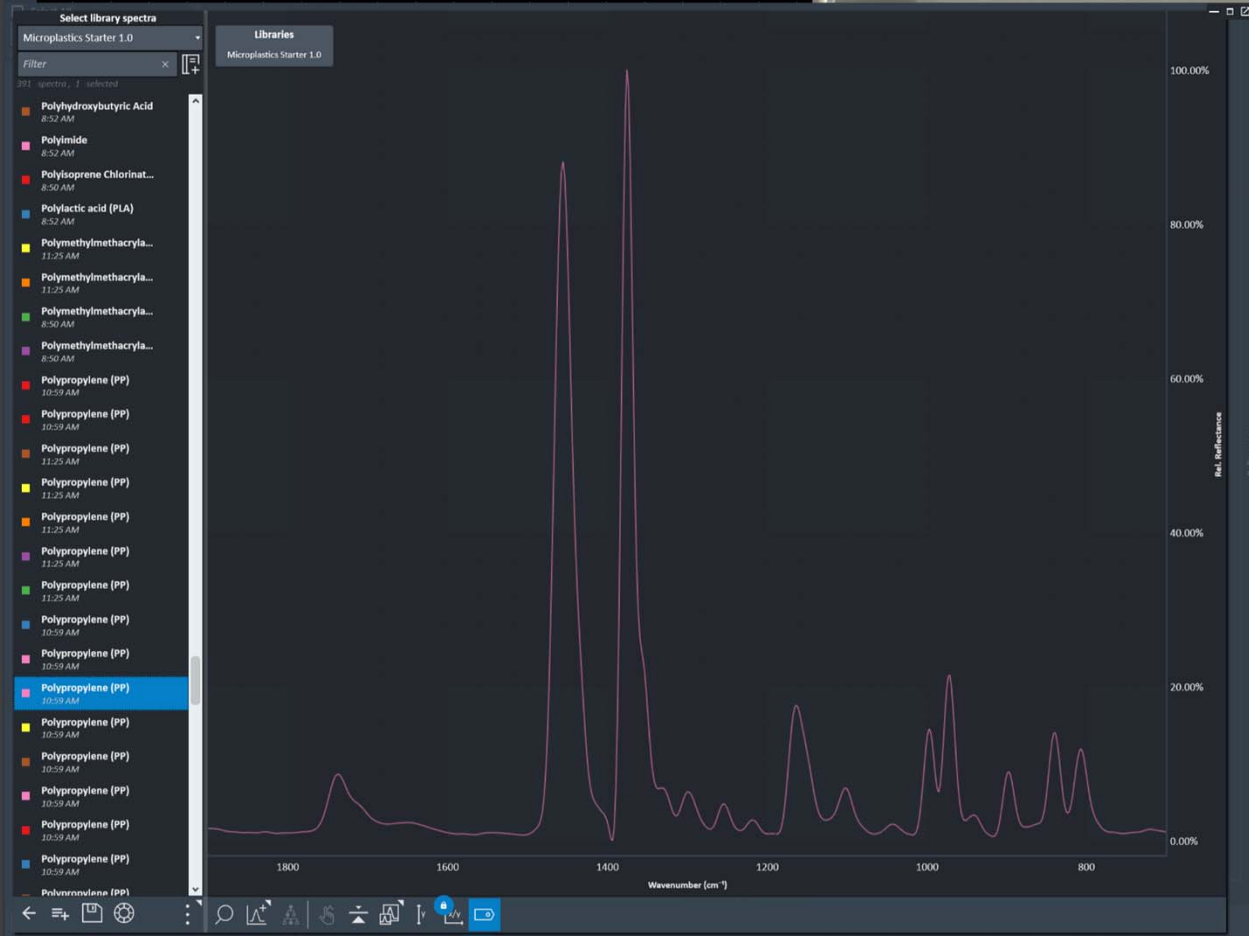
Width 810.00 µm Diameter 659.97 µm

## Reproducibility

	Sample Analyzed	LDIR Particle Count	Usable Partiles Identified	Acrylonitrile Butadiene	Alkyd Varnish	Calcium Stearate	Cellulosic	Chitin	Coal	Natural Polyamide	Polyamide (PA)	Polyethylene (PE)	Polyethylene Terephthalate (PET)	Polyimide	Polypropylene (PP)	Polytetrafluoroethylene (PTFE)	Polyurethane (PU)	Rubber	Silica
	Run 1	136	132	7	5	4	12	1	0	60	9	2	2	0	9	1	19	1	0
	Run 2	138	131	3	1	5	12	1	1	63	7	2	1	0	9	1	25	0	0
	Run 3	136	131	7	3	5	12	0	0	65	10	2	1	0	8	0	16	2	1
	Run 4	140	135	7	6	5	12	0	1	63	11	2	1	1	8	1	17	0	0
	Run 5	136	127	8	4	6	7	2	2	60	10	2	1	1	8	0	12	3	1
	Run 6	137	130	1	1	5	12	1	2	60	11	2	2	0	8	0	24	1	0
	Run 7	137	127	0	2	6	13	1	1	58	9	2	1	0	8	0	26	0	0
	Run 8	135	135	0	3	6	11	1	3	58	9	1	0	1	9	1	25	1	0
	True Number:														10				
	Mean Detected:	137	131	4.13	3.13	5.25	11.4	0.88	1.25	60.9	9.5	1.88	1.13	0.38	8.38	0.5	20.5	1	0.25
	Percent Recovery:														83.8				
	Std Dev:	1.55	3.07	3.48	1.81	0.71	1.85	0.64	1.04	2.53	1.31	0.35	0.64	0.52	0.52	0.53	5.21	1.07	0.46
	Relative Std Dev:	1.13	2.34	84.4	57.8	13.5	16.2	73.2	82.8	4.16	13.8	18.9	57	138	6.18	107	25.4	107	185



Particle Analysis



Particle Analysis

Library Microplastics Starter 1.0

Particles Identifications Statistics Settings

View All 2 particles

Search by id

Polypropylene (PP)

Quality 0.751

Id# B1  Accept Prediction

Width 221.00 µm Diameter 219.31 µm

Height 228.00 µm Area 37776.91 µm<sup>2</sup>

Wavenumber (cm<sup>-1</sup>)

Thank You!



Sidorovstock/AdobeStock